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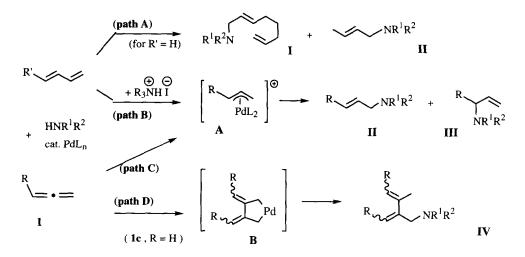
Synthesis of Allylic Amines Through the Palladium-Catalyzed Hydroamination of Allenes.

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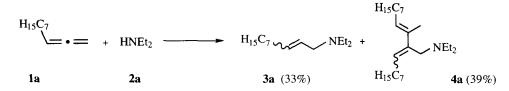
Abstract : The palladium-catalyzed addition of amines to allenes in the presence of triethylammonium iodide leads to allylic amines.

Allylic amines are important organic compounds, on account of their use as synthetic intermediates¹ and of their occurrence as natural substances.² Synthetic routes to these compounds are numerous^{3,4} and often lead to mixture of regio and stereoisomers; more recent efforts have focused on stereoselective methodologies.^{5,6} Among these, several palladium-catalyzed transformations from allylic substrates⁷, alkenes⁸, alkynes⁶ and 1,3-dienes^{8,9} have been described. Considerable work with the latter unsaturated compounds has shown that telomerization in the presence of an amine leads to octadienylamines I (1:2 adduct) or allylic amines II (1:1 adduct) when the palladium catalysts are used in association with monodentate or bidentate phosphine ligands, respectively (path A). More recently, Dieck *et al.*¹⁰ demonstrated the benificial effect of adding an amine hydroiodide salt which resulted in the formation of allylic amines II and III *via* an intermediate π -allyl complex A (path B). Allene 1c (R=H) is also known to telomerize under palladium catalysis leading to dienic amines IV (R=H), presumably through the palladacyclopentane intermediate B (path D).¹¹



We report herein on the palladium-catalyzed hydroamination of allenic compounds to substituted allylic amines II in the presence of 0.2 equivalent of triethylammonium iodide, which may involve hydropalladation¹² of the allenic functionality (path C).

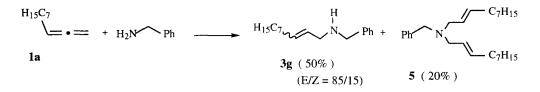
Initial experiments with 1,2-decadiene 1a and diethylamine 2a were carried out to find a suitable catalyst-ligand-solvent system for the desired transformation. Mixtures of allylic amine 3a (as a mixture E + Z) and the dienic amine 4a were obtained with palladium diacetate-triphenylphosphine in dimethylformamide (entry 1 and 2 in Table).



Other conditions met with little success (entries 3, 5) and showed that the presence of chloride ion has a detrimental effect (entries 4, 5). Finally, the catalyst bis(dibenzylideneacetone)palladium, $Pd(dba)_2$, with two equivalents of triphenylphosphine in refluxing THF allowed formation of amine **3a** and a minor amount (<10%) of the dienic amine¹³ **4a** (entries 7, 8); the donor ligand tri *n*-butylphosphine inhibited the reaction (entry 6).

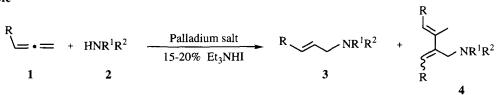
These conditions seemed general¹⁴ for the synthesis of 98% pure (E)-allylic tertiary amine **3** from a monosubstitued allene and a secondary amine (entries 7-12); allene **1c** was prone to telomerization¹¹ and gave a 55% yield of the dienic amine **4f** (entry 13). 6,7-Tridecadiene, as a model of 1,3-disubstitued allene, gave no reaction.

Another reaction with benzylamine as a primary amine and allene 1a in the same conditions gave allylic amine 3g (50%) and the bisallylic amine 5 (20%), indicating a good overall transformation of the allenic substrate.



Two plausible explanations can be proposed, each involving as the initial step formation of a hydropalladium iodide species 6. This species could hydropalladate allene 1 (path C) giving a π -allyl palladium complex 7 which is known to lead to the allylic amine 3. An alternative pathway (path E) might involve a π -complex 8 between HPdI and the terminal double bond of allene 1, which undergoes nucleophilic attack of amine 2 to give 9; reductive elimination would then afford the allylic amine 3, together with regenerated catalyst. Two similar pathways have been proposed by Yamamoto¹⁵ and our group¹⁶ for the palladium-catalyzed addition of malonate type compounds to allenes under neutral or basic conditions. However it seems presently uneasy to favour one of these pathways.¹⁵

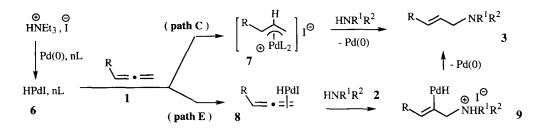
Table



Entry	Allene	Amine	Conditions	3 Yield ^a % (E/Z) ^b	4 Yield ^a %
	H ₁₅ C ₇			3a	4a
1	\=.=	HNEt ₂	$4\% Pd(OAc)_2 + 2 PPh_3$ DMF - 65°C - 19h	33 (61/39)	39
2	1a	2a	4% Pd(OAc) ₂ + 2 PPh ₃ + TMEDA DMF ~ 65°C - 18h	(16) (50/50)	(15)
3			5% Pd(PhCN) ₂ Cl ₂ + PPh ₃ MeOH - 65°C - 18h	-	-
4			4% PdCl ₂ + 4 PPh ₃ + 4% AgBF ₄ CH ₃ CN - 60°C - 20h	59 (95/5)	-
5			4% Pd(dba) ₂ + 2 PPh ₃ + 27% LiCl THF - 65°C - 18h	(38) (92/8)	(8)
6			$4\% Pd(dba)_2 + 2 P(nBu)_3$ THF - 65°C - 18h	-	-
7			5% Pd(dba) ₂ + 2 PPh ₃ THF - 65°C - 18h	55 (96/4)	9
8			$10\% Pd(dba)_2 + 2 PPh_3$ THF - 65°C - 18h	70 (96/4)	8
9	1a	€ ^{NH} 2b	5% Pd(dba) ₂ + 2 PPh ₃ THF - 65°C - 17-22h	3b 53 (98/2)	-
10			"	3c 64 (98/2)	4c 14
11	Ph	Et ₂ NH 2a	11	3d 72 (98/2)	4d 19
12	1b	€ ^{NH} 2b	"	3e 89 (98/2)	-
13	$H_2C = \bullet = CH_2$ 1c	С ^{NН} 2b	"	-	4f 55

a : Yields refer to materials isolated by column chromatography, except those in parentheses which are determined through GLC (hexadecane as internal standard).

b: E/Z ratios were determined through GLC or ¹H NMR



Further studies on the mechanistic aspects of this new hydroamination reaction of allenes are in progress, as well as synthetic application to functionalized allenes, e.g. allenic alcohols. Thus 1,2-nonadien-4-ol **10** gave a 60% yield of the allylic aminoalcohol **11** under the above described conditions.



References and notes :

- For recent uses in organic synthesis, see for example : Burgess, K.; Ohlmeyer, M.J. J. Org. Chem. 1991, 56, 1027-1036. Jumnah, R.; Williams, J.M.J.; Williams, A.C. Tetrahedron Lett.. 1994, 34, 6619-6622.
- 2. Deleris, G; Dunogues, J.; Gadras, A. Tetrahedron 1988, 44, 4243-4258, and refences cited therein Stütz, A. Angew. Chem., Int. Ed. Engl. 1987, 26, 320.
- 3. For a review, see : Cheikh, R.B. ; Chaabouni, R. ; Laurent, A. ; Mison, P. ; Nafti, A. Synthesis 1983, 685-699.
- 4. Corriu, R.J.P.; Geng, B.; Moreau, J.J.E. J. Org. Chem. 1983, 58, 1443-1448.
- 5. Capella, L.; Degl'innocenti, A.; Mordini, A.; Reginato, G.; Ricci, A.; Seconi, G. Synthesis **1991**, 1201-1204.
- 6. Corriu, R.J.P.; Bolin, G.; Moreau, J.J.E. Bull. Soc. Chim. Fr. 1993, 130, 273-280.
- Trost, B.M.; Keinan, E. J. Am. Chem. Soc. 1978, 100, 7779-7781. Trost, B.M.; Keinan, E. J. Org. Chem. 1978, 44, 3451-3457. Mitsunobu, O. In Comprehensive Organic Synthesis; Trost, B.M. and Fleming, I., Eds.; Pergamon Press: Oxford, 1991, Vol 6, p. 65.
- 8. Takahashi, K.; Miyake, A.; Hata, G. Bull. Chem. Soc. Jpn. 1972, 45, 1183-1191.
- Keim, W.; Behr, A.; Röper, M. In Comprehensive Organometallic Chemistry; Wilkinson, G.; Stone, F.G.; Abel, E.W., Eds.; Pergamon Press: Oxford, Vol 8, 1983, p. 371.
- 10. Armbruster, R.W.; Morgan, M.M.; Schmidt, J.L.; Lau, C.M.; Riley, R.M.; Zabrowski, D.L.; Dieck, H.A. Organometallics 1986, 5, 234-237.
- 11. Coulson, R. J. Org. Chem. 1973, 38, 1483-1490.
- 12. A hydropalladation step has been postulated in the Pd-catalyzed addition of an acid function to propargylic acetate to give allylic gem-dicarboxylate :
 - Trost, B.M.; Brieden, W.; Baringhaus, K.H. Angew. Chem. Int. Ed. Engl. 1992, 31, 1335-1336.
- 13. In all the cases the dienic amine 4a, the product of the telomerization-functionalization of allene 1a, was obtained as a 70/30 mixture of two of its stereoisomers which were not separated by flash-chromatography.
- 14. Typical procedure is as follows : N-(3-phenyl-2-propenyl) pyrrolidine **3e** (entry 12) Pd(dba)₂ (0.13mmol), triphenylphosphine (0.26 mmol) and Et₃NHI (0.44 mmol) were mixed in THF (3 ml) under nitrogen during 15 minutes. Phenylallene **1b** (2.6 mmol) and pyrrolidine **2b** (2.88 mmol, 1.1 equiv) were then added and the mixture heated to 60°C for 32h. Hydrolysis with water (10 ml) was followed by extraction with CH₂Cl₂ (4x15 ml) and drying over Na₂SO₄. Purification by flash-chromatography on silicagel (CH₂Cl₂/MeOH : 95/5 + 0.1% NH₄OH) yielded 190 mg (89%) of amine **3e**. ¹H NMR (200 MHz, CDCl₃), δ (ppm) : 1.05 (t, J = 7.1 Hz, 6H) ; 2.56 (q, J = 7.1 Hz, 4H) ; 3.23 (d, J = 6.5 Hz, 2H) ; 6.29 (dt, J_{trans} = 15.9 Hz and J = 6.5 Hz, 1H) ; 6.50 (d, J_{trans} = 15.9 Hz, 1H) ; 7.14-7.38 (m, 5H).
- 15. Yamamoto, Y.; Al-Massum, M.; Asao, N. J. Am. Chem. Soc. 1994, 116, 6019-6020.
- 16. Besson, L.; Goré, J.; Cazes, B. Tetrahedron Lett. 1995, submitted, previous note in this issue.