by means of lithium aluminum hydride to the base III in high yield. The product was stable to distillation

and was converted into the hydrochloride and methiodide salts.

The dihydrochloride of base III (R = dimethylaminopropyl) when screened against KB tissue culture cells was active at about 10  $\mu g/ml$ . None of the compounds showed any activity against L1210 lymphoid leukemia. The dimethiodide of III (R = dimethylaminopropyl) produced ganglionic blockage when tested on the nictitating membrane of the cat and gave a moderate reduction of blood pressure in an anesthetized dog.

## Experimental Section<sup>5</sup>

N-(3-Dimethylaminopropyl)-1,2,3,4,5,6,7,8,8a,9,10,10a-dodecahydro-9,10-phenanthrenedicarboximide (II, <math display="inline">R= dimethylaminopropyl).—To 5 g (0.0192 mole) of finely powdered anhydride I was added, with shaking, 2.5 g (excess) of 3-dimethylaminopropylamine. After the initial reaction the mixture was heated at 180–200° for 30 min. The product distilled as a viscous glass, bp 210–220° (0.07 mm), yield 5 g (76\%). Anal. (C21H32N2O2) C, H, N.

The monomethiodide, prepared in the usual manner, melted at 208–210°. Anal. ( $C_{22}H_{35}IN_2O_2$ ) I.

N-(3-Dimethylaminopropyl)-3a,3b,4,5,6,7,8,9,10,11,11a,11b-dodecahydrodibenz[e,g] isoindoline (III, R = dimethylaminopropyl).—To a solution of 9 g of LiAlH<sub>4</sub> hydride in 1 l. of anhydrous ether, the imide II dissolved in 1 l. of anhydrous ether was added rapidly with vigorous stirring and the mixture was refluxed 4 hr. While stirring, the reaction mixture was decomposed by the dropwise addition of H<sub>2</sub>O (36 ml) and the stirring was continued for an additional 3 hr. After standing overnight the solution was filtered and the inorganic cake was washed (dry Er<sub>2</sub>O). The solution was dried (Na<sub>2</sub>SO<sub>4</sub>), the solvent was removed, and the residual oil was distilled, bp 160–170° (0.07 mm). The distillate weighed 3.3 g (80%). Anal. (C<sub>21</sub>H<sub>3e</sub>N<sub>2</sub>) C, H, N.

The dihydrochloride prepared in the usual manner after recrystallization from EtOH-Me<sub>2</sub>CO melted at 305-306° (put in bath at  $290^{\circ}$ ). Anal. ( $C_{21}H_{38}Cl_2N_2$ ) Cl, N.

The dimethiodide was prepared in EtOH by refluxing with excess MeI and diluting with 3 vol. of EtOAc. After recrystallization from EtOH-Me<sub>2</sub>O with a trace of ether the crystals melted at 240-242° dec. Anal. (C<sub>23</sub>H<sub>42</sub>I<sub>2</sub>N<sub>2</sub>) I, N.

N-(2-Dimethylaminoethyl)-3a,3b,4,5,6,7,8,9,10,11,11a,11b-dodecahydrodibenz [e,g] isoindoline (III,  $\mathbf{R}=$  dimethylaminoethyl) was prepared as outlined above except that the imide was not distilled. The crude imide was dissolved in ether and reduced (LiAlH<sub>4</sub>, bp 155–160° (0.05 mm). Anal. ( $C_{20}H_{34}N_2$ ) C, H, N.

The dihydrochloride, prepared in the usual way, melted at  $259-260^{\circ}$  dec. Anal. ( $C_{20}H_{36}Cl_2N_2$ ) Cl.

The dimethiode prepared as described above melted at 241–242° dec. Anal.  $(C_{22}H_{40}I_2N_2)$  I.

N-(3-Morpholinopropyl)-3a,3b,4,5,6,7,8,9,10,11,11a,11b-dodecahydrodibenz [e,g] isoindoline (III, R = morpholinopropyl) was prepared as above from 5 g of the anhydride without isolation of the imide. It boiled at  $180-190^{\circ}$  (0.07 mm) and weighed 4.1 g. Anal. ( $C_{23}H_{38}N_2O$ ) C, H, N.

The dihydrochloride when recrystallized from EtOH-Et<sub>2</sub>O melted at 286-289°. Anal. (C<sub>23</sub>H<sub>40</sub>Cl<sub>2</sub>N<sub>2</sub>O) Cl, N.

The dimethiodide was prepared in MeOH and precipitated with EtOAc. When recrystallized from EtOH-ether it melted at  $246-248^{\circ}$ . Anal. (C<sub>2</sub>,H<sub>4</sub>I<sub>2</sub>N<sub>2</sub>O) I, N.

(5) Melting points were determined with a Thomas-Hoover apparatus and are corrected. Microanalyses were performed by Schwarzkopf Microanalytical Laboratory, Woodside, N. V. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within 0.3% of the theoretical valves.

## The Ethyl Homologs of 2,4,5-Trimethoxyphenylisopropylamine

ALEXANDER T. SHULGIN

1483 Shulgin Road, Lafayette, California

Revised Manuscript Received August 23, 1967 Revised Manuscript Received August 23, 1967

Of the six possible 1-(trimethoxyphenyl)-2-aminopropanes (trimethoxyamphetamines), the 2,4,5 isomer (IIa) was the most potent as a psychotomimetic agent, and it serves in this present report as the reason for the synthesis of the seven possible ethyl homologs. These have been prepared by routes which preclude isomer contamination. In preliminary observations only the 4-monocthoxy isomer IIe exceeds IIa in psychotomimetic potency.

## **Experimental Section**

The 2-ethoxy homolog (Hb) was prepared by the Claisen rearrangement of allyl 3,4-dimethoxyphenyl ether as described earlier. The remaining isomers (Hc-h) employed the three separate 3,4-dialkoxyphenols obtained by the peracetic acid oxidation of the appropiate aldehyde. The synthesis of 2,4-dimethoxy-5-ethoxyphenylisopropylamine (He) is typical. The malononitrile derivatives were prepared as described earlier. The microanalyses of all new compounds in Tables I and II are listed in Table III; melting points were determined on a Kofler Heizbank and are corrected.

$$\begin{array}{c} \text{Table I} \\ \\ R^2 \\ \end{array} \begin{array}{c} \text{CHO} \end{array}$$

			~= ···-·Mp, ∘C····· · ∨			
RI	$\mathbf{R}^{z}$	$\mathbb{R}^3$	ArCHO	ΛrCH== C(CN) <sub>2</sub>		
H	$OCH_3$	$\mathrm{OCH}_3$	$45^a$	$147^{b}$		
H	$\rm OCH^3$	$\mathrm{OC_2H_5}$	$50^{\circ}$	142		
H	$\mathrm{OC_2H_5}$	$OCH_3$	$60^d$	141		
H	$\mathrm{OC_2H_5}$	$\mathrm{OC_2H_5}$	$\mathrm{Oil}^e$	1057		
$OCH_3$	$OCH_3$	$OC_2H_5$	108#	136		
${\rm OCH_3}$	$\mathrm{OC_2H_5}$	$OCH_3$	$109^{h}$	172		
$OCH_3$	$\mathrm{OC_2H_5}$	$\mathrm{OC_2H_5}$	89	157		
$\mathrm{OC_2H_5}$	$OCH_3$	$\mathrm{OC_2H_5}$	111	158		
$\mathrm{OC_2H_5}$	$OC_2H_5$	${\rm OCH_3}$	99	173		
$OC_2H_5$	$OC_2H_5$	$OC_2H_3$	$95^i$	170		

<sup>a</sup> L. Gattermann [Ann., **357**, 313 (1907)] reported mp 43-44°. <sup>b</sup> H. Kauffmann [Ber., **52**, 1422 (1919)] reported mp 147°. <sup>c</sup> E. Spath and E. Bernhauer [ibid., **58**, 200 (1925)] reported mp 150-151°. <sup>d</sup> F. Tiemann [ibid., **8**, 1127 (1875)] reported mp 64-65°. <sup>e</sup> Obtained from the Eastman Kodak Co. <sup>f</sup> R. P. Mariella and J. M. Bauer [J. Org. Chem., **23**, 120 (1958)] reported mp 104-104.5°. <sup>g</sup> F. S. H. Head and A. Robertson [J. Chem. Soc., 2434 (1930)] reported mp 110°. <sup>h</sup> Lit.<sup>g</sup> mp 110°. <sup>i</sup> W. Will [Ber., **16**, 2106 (1883)] reported mp 95°.

1,3-Dimethoxy-4-ethoxybenzene.—To a solution of 4-ethoxy-3-methoxyphenol in MeOH (14 g in 20 ml) was added a solution of 5.3 g of KOH in MeOH (100 ml), followed by MeI (11.9 g). The mixture was refluxed for 2 hr, quenched with 3 vol of  $\rm H_2O$ , and made strongly basic with 5% NaOH. Extraction with ether and evaporation of the pooled extracts yielded the title ether as a clear oil, 9.7 g,  $n^{25}$ p 1.5210.

<sup>(1)</sup> A. T. Shulgin, J. Med. Chem., 9, 445 (1966).

<sup>(2)</sup> A. T. Shulgin, Experientia, 20, 366 (1964).

<sup>(3)</sup> C. Naranjo, T. Sargent, and A. T. Shulgin, unpublished data

TABLE II

$$R^{2}$$
 $CH = CNO_{2}$ 
 $R^{2}$ 
 $CH_{3}$ 
 $R^{3}$ 
 $CH_{3}$ 
 $CH_{3}$ 
 $R^{3}$ 
 $CH_{3}$ 
 $CH_{3}$ 

	$\mathbb{R}^1$	$\mathbb{R}^2$	$\mathbb{R}^3$	Mp, °C	Color	Yield, %	Mp, °C	Yield, %	
а	$\rm OCH_3$	$OCH_3$	$OCH_3$	102	Yellow	83	181	92	
b	$\mathrm{OC}_2\mathrm{H}_{\ddot{\epsilon}}$	$OCH_3$	$OCH_3$	76	Yellow	73	172	67	
$\mathbf{c}$	$OCH_3$	$OCH_3$	$\mathrm{OC_2H_5}$	97	Yellow	27	172	74	
$\mathbf{d}$	$OC_2H_5$	$\mathrm{OCH}_3$	$\mathrm{OC_2H_5}$	86	Orange	36	164	76	
e	$OCH_3$	$\mathrm{OC_2H_5}$	$OCH_3$	129	Orange	52	172	70	
f	$OC_2H_5$	$\mathrm{OC}_2\mathrm{H}_5$	$OCH_3$	113	Orange	72	159	62	
g	$OCH_3$	$\mathrm{OC_2H_5}$	$OC_2H_5$	91	Orange	38	180	75	
${f h}$	$OC_2H_5$	$\mathrm{OC_2H_5}$	$\mathrm{OC_2H_5}$	92	Yellow	40	168	74	

TABLE III ANALYSES

	111	TIMETOES					
Compound	Formula	C	—Calcd, %— H	N	C	-Found, %- H	N
·	r of muta	O	11	24	C	11	
Benzaldehyde	C H $O$	04.07	7 10		04.4	<b>-</b> 0	
4,5-Diethoxy-2-methoxy-	$\mathrm{C_{12}H_{16}O_{4}}$	64.27	7.19	• • •	64.4	7.0	• • •
2,5-Diethoxy-4-methoxy-					64.5	7.1	
2,4-Diethoxy-5-methoxy-					64.3	7.3	
Benzylidenemalononitrile	G II N O	40.41	<b>*</b> 00	40.0	20.2	• 0	10.1
5-Ethoxy-4-methoxy-	$\mathrm{C_{13}H_{12}N_{2}O_{2}}$	68.41	5.30	12.27	68.2	5.2	12.1
4-Ethoxy-5-methoxy-	~ ** ** *				68.2	5.1	11.9
2,4-Dimethoxy-5-ethoxy-	$\mathrm{C_{14}H_{14}N_{2}O_{3}}$	65.11	5.46	10.85	65.1	5.4	10.7
2,5-Dimethoxy-4-ethoxy-					64.9	5.2	10.7
4,5 Diethoxy-2-methoxy-	$\mathrm{C_{15}H_{16}N_{2}O_{3}}$	66.16	5.92	10.29	66.0	5.9	10.1
2,5-Diethoxy-4-methoxy-					66.0	5.7	10.3
2,4-Diethoxy-5-methoxy-					65.9	5.7	10.1
2,4,5-Triethoxy-	${ m C_{16}H_{18}N_2O_3}$	67.12	6.34	9.78	67.1	6.3	9.6
Phenylnitropropene							
2-Ethoxy-4,5-dimethoxy-	${ m C_{13}H_{17}NO_{5}}$	58.42	6.41	5.24	58.3	6.5	5.2
5-Ethoxy-2,4-dimethoxy-					58.3	6.3	5.1
4-Ethoxy-2,5-dimethoxy-					58.2	6.3	5.2
2,5-Diethoxy-4-methoxy-	$\mathrm{C_{14}H_{19}NO_{5}}$	59.77	6.81	4.98	59.7	6.7	5.0
2,4-Diethoxy-5-methoxy-					59.5	6.7	4.8
4,5-Diethoxy-2-methoxy-					59.5	6.7	4.9
2,4,5-Triethoxy-	$\mathrm{C}_{15}\mathrm{H}_{21}\mathrm{NO}_{5}$	61.00	7.17	4.74	60.9	7.1	4.6
Amphetamine hydrochloride							
2-Ethoxy-4,5-dimethoxy-	$\mathrm{C_{13}H_{22}NO_{3}Cl}$	56.62	8.04	5.08	56.3	7.9	5.0
4-Ethoxy-2,5-dimethoxy-					56.2	7.9	5.0
5-Ethoxy-2,4-dimethoxy-a					55.9	7.9	4.9
y =y					55.9	7.9	5.0
2,4-Diethoxy-5-methoxy-	$\mathrm{C_{14}H_{24}NO_{3}Cl}$	58.02	8.35	4.83	57.7	8.3	$\frac{3.5}{4.7}$
2,5-Diethoxy-4-methoxy-a	01411241 0801	00.02	0.00	1.00	57.4	8.3	4.7
4,5-Diethoxy-2-methoxy-					57.4 57.6	8.3	4.7
2,4,5-Triethoxy-	$\mathrm{C_{15}H_{26}NO_{3}Cl}$	59.30	8.63	4.61	$57.0 \\ 59.2$	8.6	4.5
L, I, O I I I COHONY	013112011 0301		0.00	T.U1	00.4	0.0	±.0

<sup>&</sup>lt;sup>a</sup> These had rather low carbon values and were both analyzed by nmr (A-60, in D<sub>2</sub>O) and both appeared to be without contaminants.

2,4-Dimethoxy-5-ethoxybenzaldehyde.—A mixture of N-methylformanilide and POCl<sub>3</sub> (17.3 g in 19.6 g) was allowed to stand at room temperature for 0.5 hr. There was then added 9.2 g of the above ether and this mixture was heated for 2 hr on the steam bath. The resulting black viscous product was poured onto 800 ml of cracked ice and allowed to stand overnight. The crude aldehyde was removed by filtration and was recrystallized from 100 ml of MeOH to yield 8.8 g of fluffy white crystals.

The crude reaction product of this synthesis, as well as of each of the other five parallel Vilsmeyer preparations, was subjected to glpc analysis. Two separate substrate systems were employed, a  $300 \times 1$  cm column containing 5% 710 Silicone on 60–80 firebrick and a  $150 \times 1$  cm column with 15% ethylene glycol succinate on 60-80 acid-washed Chromosorb W. In no case was an isomeric aldehyde evident in excess of 2% although N-methylformanilide consistently appeared to the extent of several per cent. The recrystallization step in every case effectively removed these contaminants.

1-(2,4-Dimethoxy-5-ethoxyphenyl)-2-nitropropene (Ic).—A solution of 2,4-dimethoxy-5-ethoxybenzaldehyde in AcOH (6.7 g in 25 g) was treated with 2.1 g of NH<sub>4</sub>OAc followed by 3.3 g of nitroethane. The mixture was heated on the steam bath for 2 hr. After cooling, the addition of a small amount of water caused the deposition of the product as a thick gel which was separated by recrystallization from toluene. As noted in Table II, where both the yields and the physical properties of these nitropropenes are shown, all compounds with a 5-ethoxy group were obtained in poor yields in this nitroethane coupling step.

2,4-Dimethoxy-5-ethoxyphenylisopropylamine (IIc).—The nitrostyrene Ic, as well as all others in this study, was reduced by the Soxhlet technique employed by Ramirez and Burger<sup>4</sup> with the work-up modification described earlier.<sup>1</sup> Again, the yields and properties are recorded in Table II.

<sup>(4)</sup> F. A. Ramirez and A. Burger, J. Am. Chem. Soc., 72, 2782 (1950).